



GOVT. RAJMATA VIJAYA RAJE SINDHIYA KANYA MAHAVIDYALAYA, KAWARDHA
Kawardha, Dist:- Kabirdham, Chhattisgarh
Registered Under Section 2(F) of UGC Act
Affiliated to Hemchand Yadav University, Durg (C.G.)
Accredited 'B' Grade by NAAC



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S. NO...../Seminar/2025 Kawardha

Date .08/11/2025

Seminar notice

It is notify to students of M.Sc. I & III Chemistry Semester for presenting ppt/other in Class seminar. It is a part of skill developments of students. Hence, the students will participate in seminar. The seminar will be conduct on 08/11/2025 from time 11:30am . The lists of participated students are below here.

S.No.	Name of Students	Topic	Remark
1.	Dhaneshwari M.Sc.III rd	Thermogravimetric Analysis	
2.	Tuleshwari Jangde M.Sc.I st	Aromaticity	
3.	Hemlata Yadav M.Sc.I st	Crystal Field Theory	


Dr. Birendra Kumar
Head of Department

Chemistry
HOD (Chemistry)
Govt. Rajmata Vijayaraje Sindhiya
Kanya Mahavidyalaya, Kawardha
Distt.-Kabirdham (C.G.)


Principal

PRINCIPAL
New Govt. Girls College
Kawardha
Distt.-Kabirdham (C.G.)

1st Seminar 08/11/2025

MSc. Chemistry IIIrd sem and 7th sem.

Date - 08/11/2025

Time - 11.30 am

Today Seminar presented by
Tulshurary MSc. 7th sem. her topic is "Aromaticity"
and hemlata Yadav msc 7th sem. her topic is
"Crystal Field Theory" and Phaneshwar Kaushik
MSc. IIIrd sem. her topic is "TGA".

This session is very knowledgeable
for us in this seminar following student
are present.

S.No.	Name of student	Class	Sig.
①	Roshani Kaushik	III rd sem.	(R)
②	Bina Sahu	III rd sem.	(B)
③	Rusamangli Patre	III rd sem.	(R)
④	Phaneshwar Kaushik	III rd sem.	(P)
⑤	Purnima Sahu	I sem	(P)
⑥	Premeshwari	I sem	(P)
⑦	Gamini Satyawanshi	I sem	(S)
⑧	Roshni Dindore	I sem	(R)
⑨	Sanika Patre	I sem.	(S)
⑩	Neelam Gendre	I sem.	(N)
⑪	Santoshi Kaushik	I sem.	(S)
⑫	Nitya Kaushik	I sem.	(N)

⑬	Hemlata Yadav	I sem.	(H)
⑭	Vansha Upadhyay	I st sem	(V)
⑮	Tulshurary Jangde	I st sem.	(T)
⑯	Nisha Manikpuri	III sem	(N)

Sahu
08/11/25

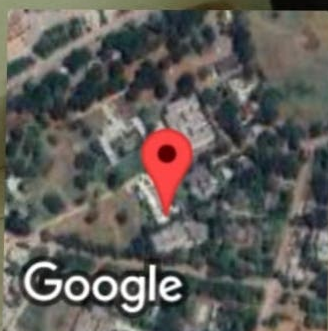
Dr. Birendra Kumar

Sanghal
08/11/25

Govt. Rajm...
Kanya Mahavidyalaya Kawardha
Distt. Kabirdham (C.G.)



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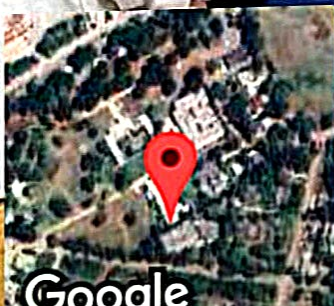
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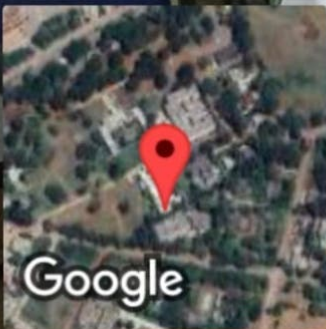
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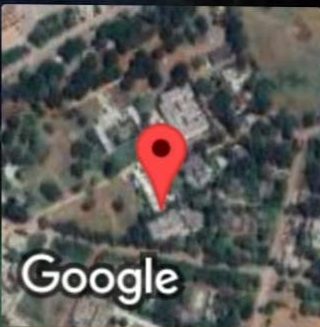
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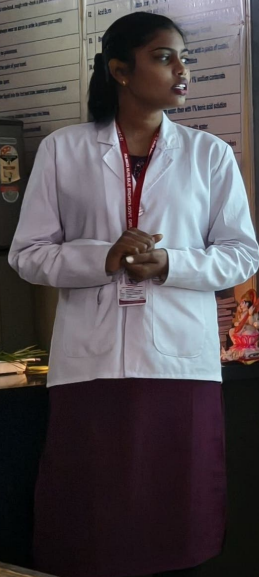
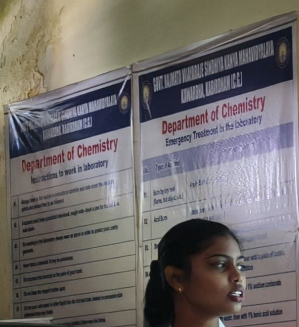
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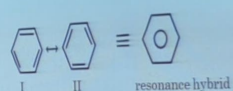


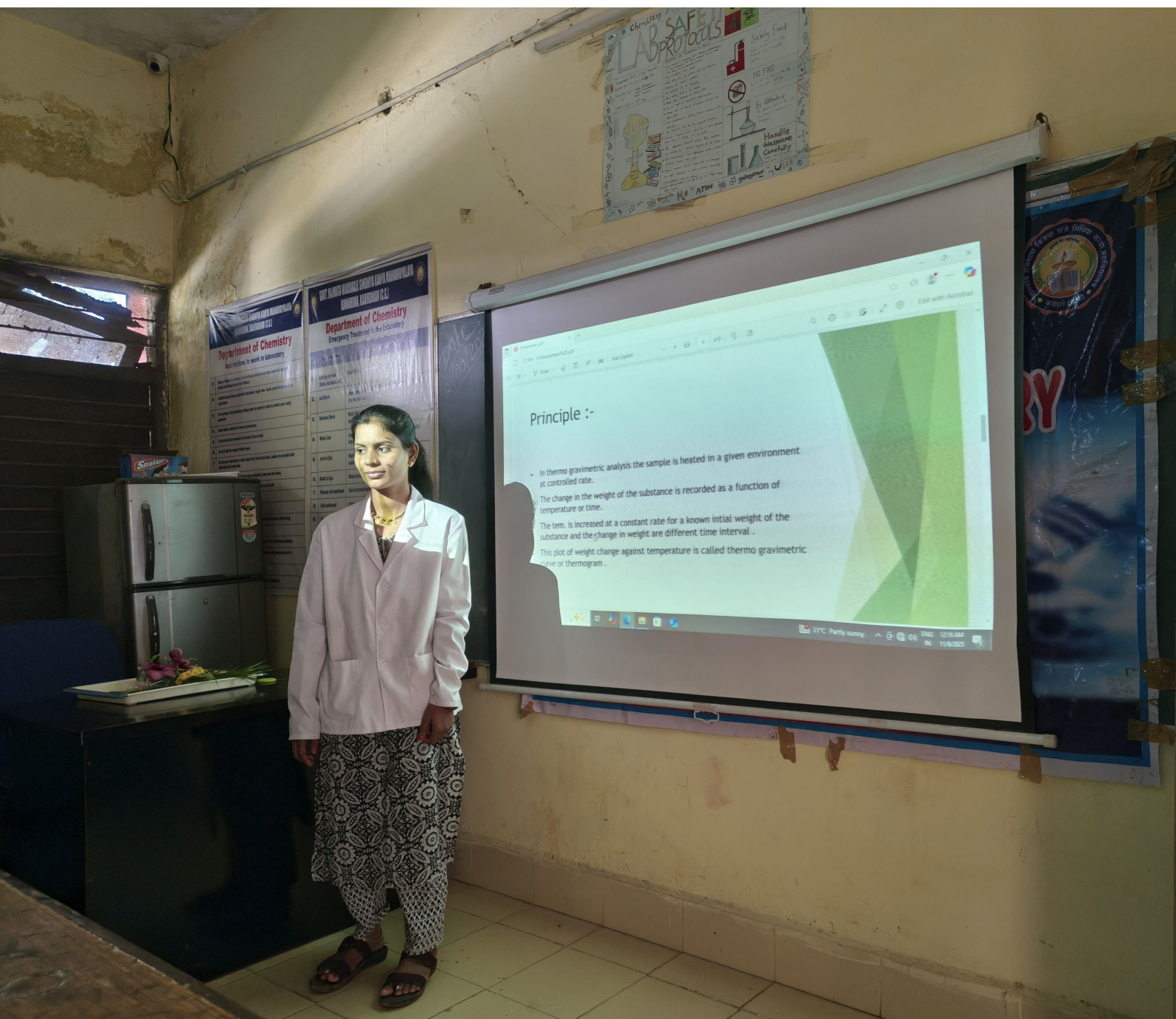


AROMATICITY IN BENZENOID COMPOUND

➤ Benzenoid aromatic compounds are the organic molecular species either with isolated benzene ring or with multiple benzene ring which fused to form a more complex structure.

➤ Aromatic molecules are very stable and generally undergo electrophilic substitutions rather than addition. Since the most common aromatic compound are derivatives of benzene



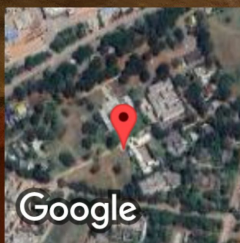


Principle :-

- In thermo gravimetric analysis the sample is heated in a given environment at controlled rate.
- The change in the weight of the substance is recorded as a function of temperature or time.
- The tem. is increased at a constant rate for a known initial weight of the substance and the change in weight are different time interval .
- This plot of weight change against temperature is called thermo gravimetric curve or thermogram .



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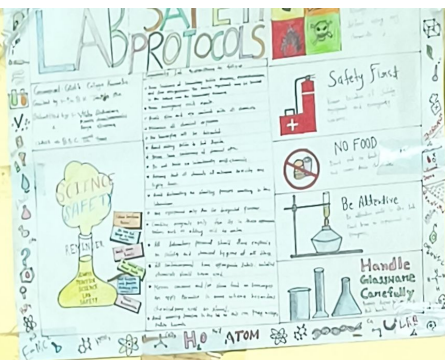
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